



IFM-GEOMAR

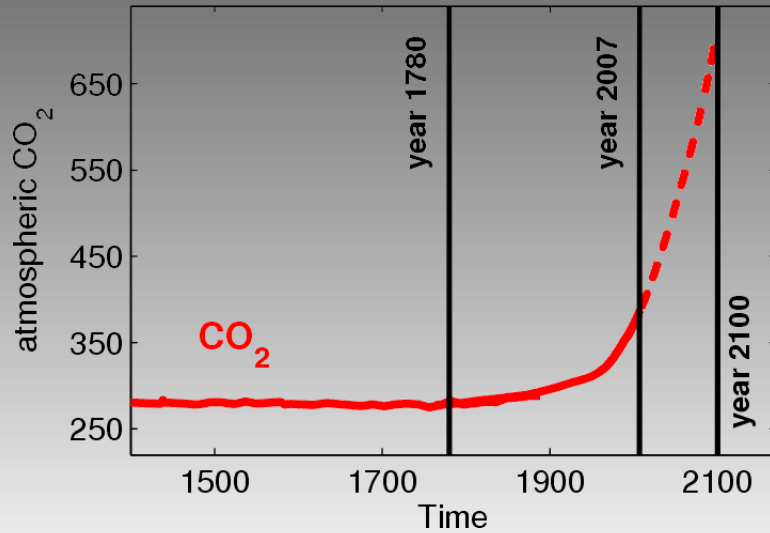
Leibniz-Institut für Meereswissenschaften
an der Universität Kiel

Experimental approaches of carbonate chemistry manipulation in CO₂ perturbation studies

K. G. Schulz, U. Riebesell

Monaco, 06.-09.10.2008

Atmospheric CO₂ variability and marine life

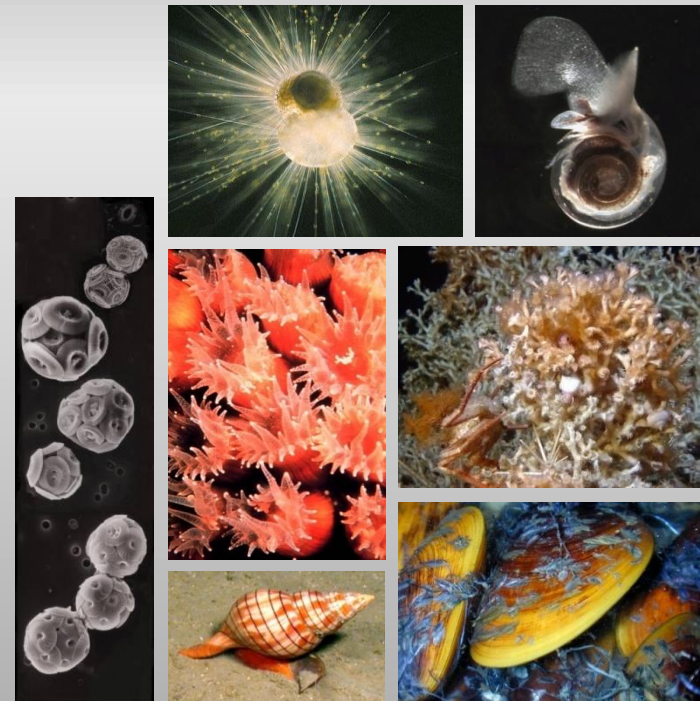


increase to about 700 ppmv until the year 2100 (IS92a)

100 ppmv increase since industrial revolution

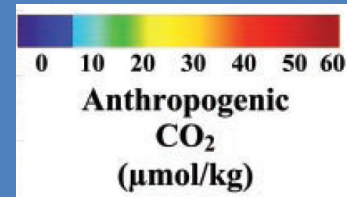
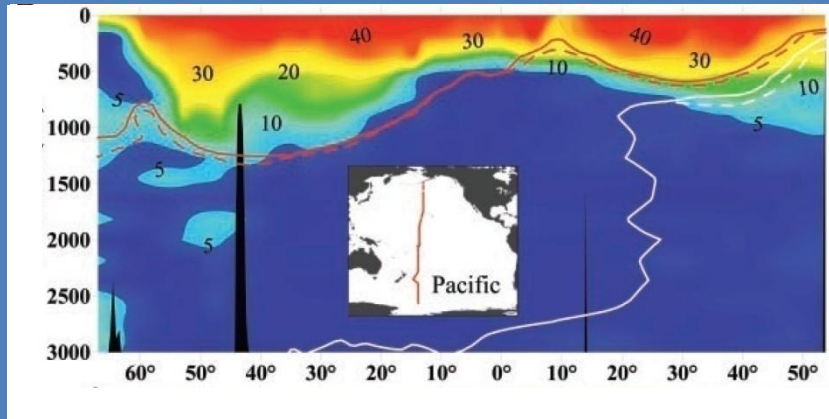
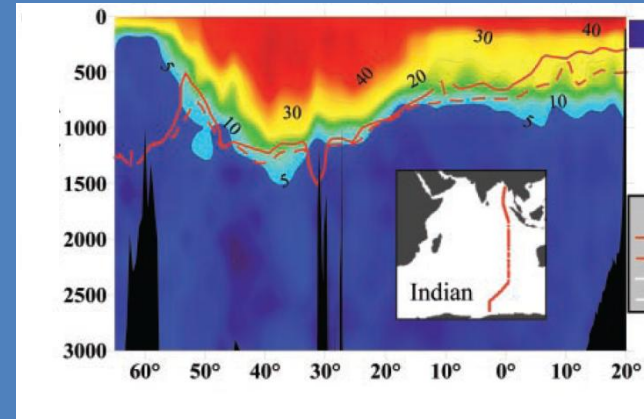
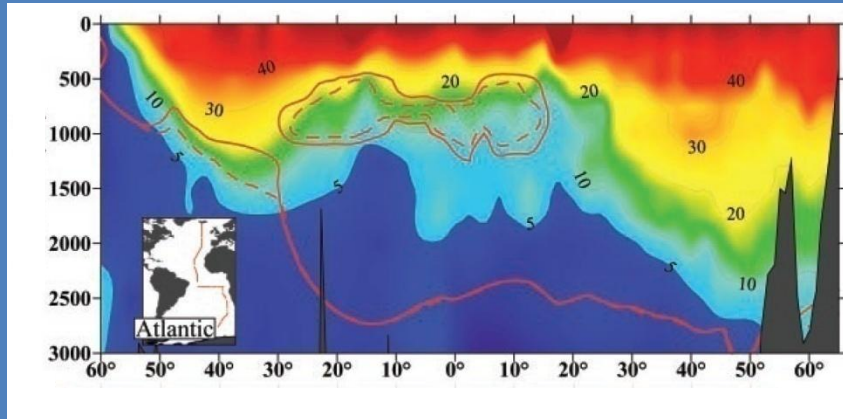
Siegenthaler et al. 2005, Dr. Peter Tans (NOAA/ESRL), IPCC 2001

➔ especially of interest for marine calcifiers, such as mussels, snails, Corals, foraminifera, pteropods or coccolithophorids



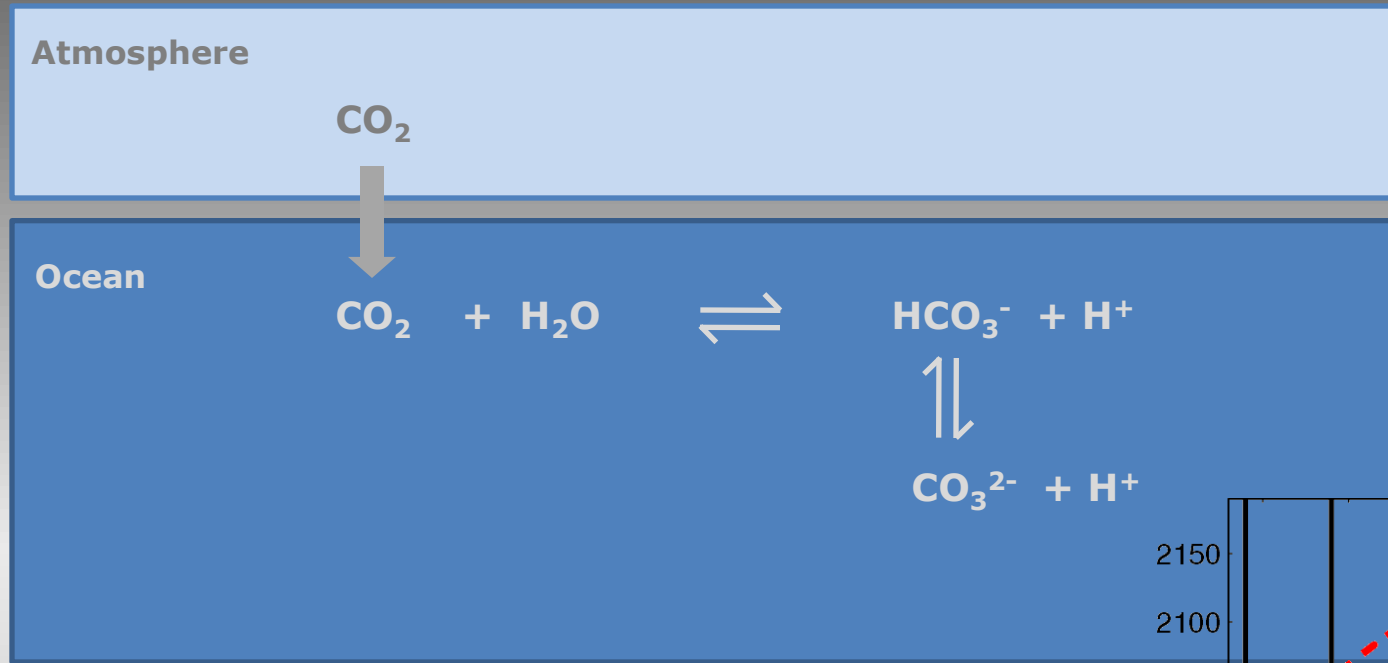
Anthropogenic CO₂ invasion until present day

Depth (m)



Latitude

CO₂ dissociation in seawater

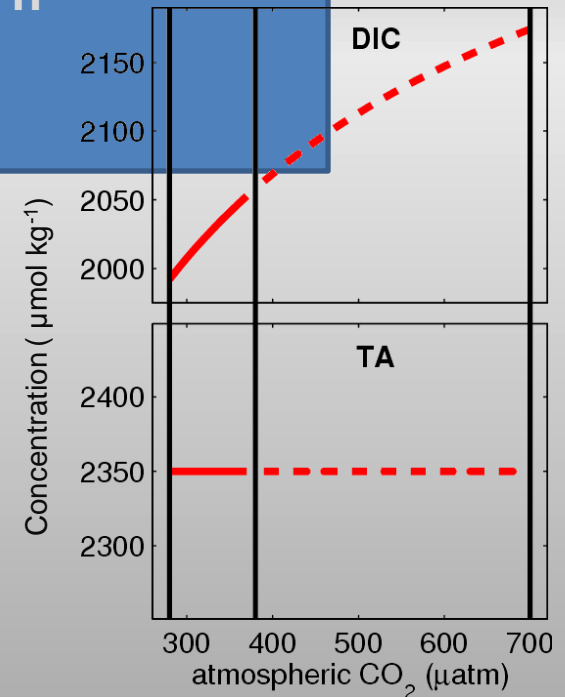


dissolved inorganic carbon

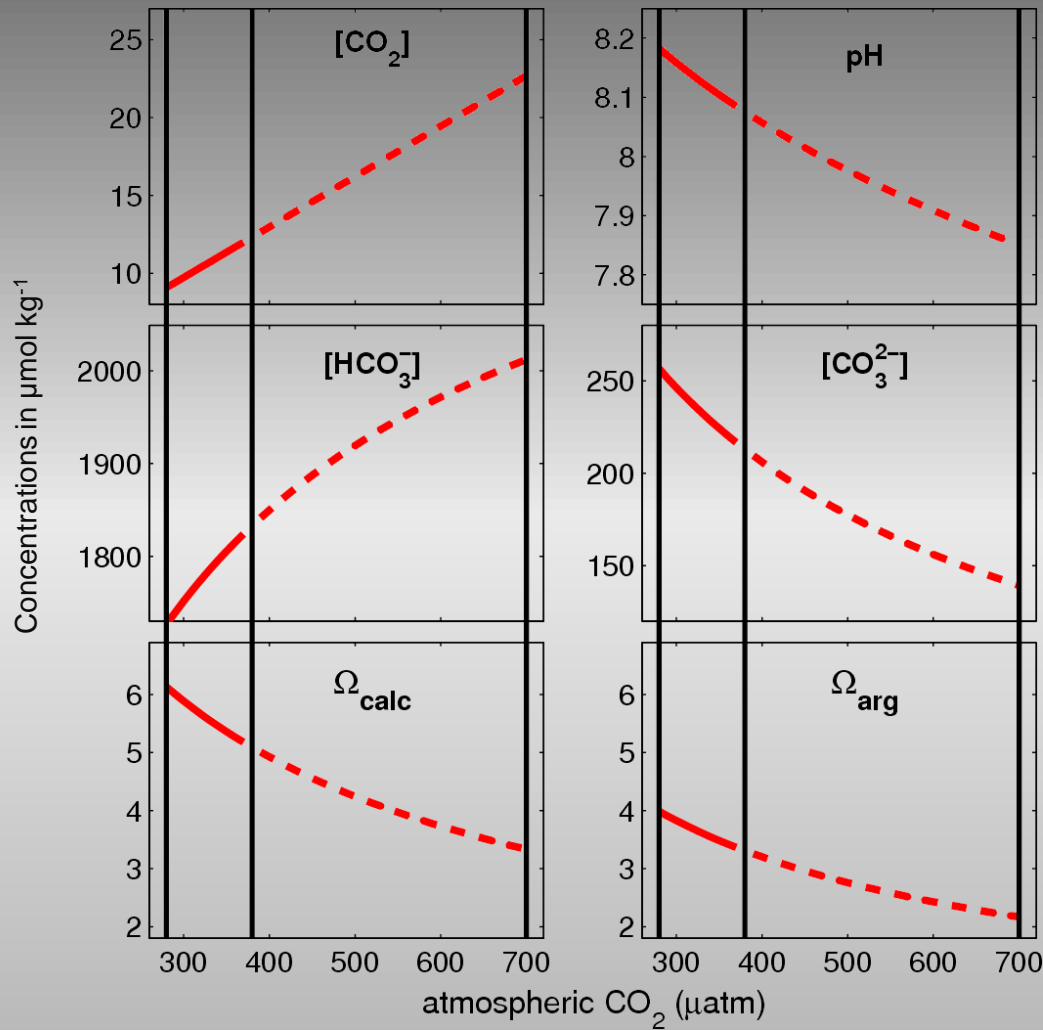
$$\text{DIC} = [\text{CO}_2] + [\text{HCO}_3^-] + [\text{CO}_3^{2-}]$$

total alkalinity (excluding some minor components)

$$\text{TA} = [\text{HCO}_3^-] + 2[\text{CO}_3^{2-}] + [\text{B}(\text{OH})_4^-] + [\text{OH}^-] - [\text{H}^+]$$



Surface seawater carbonate chemistry redistribution



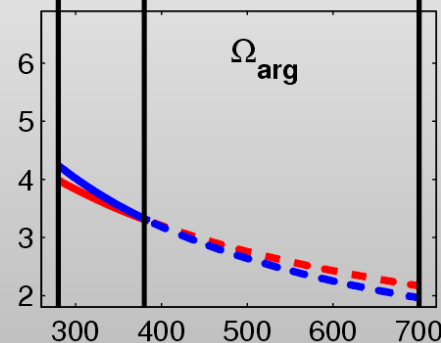
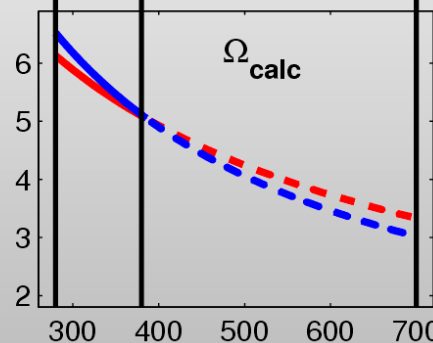
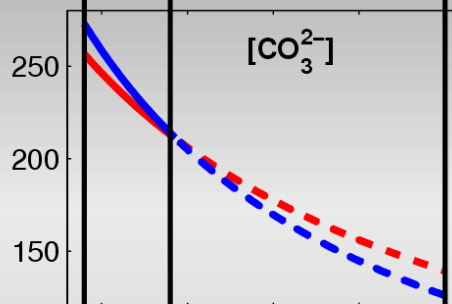
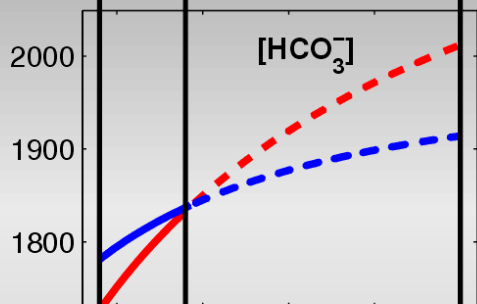
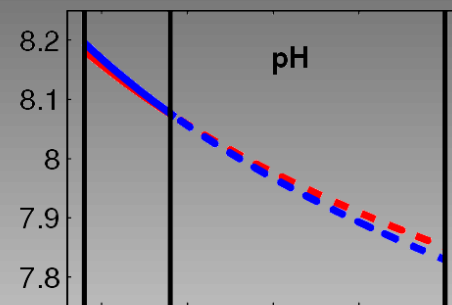
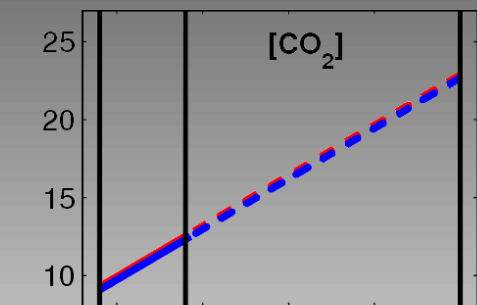
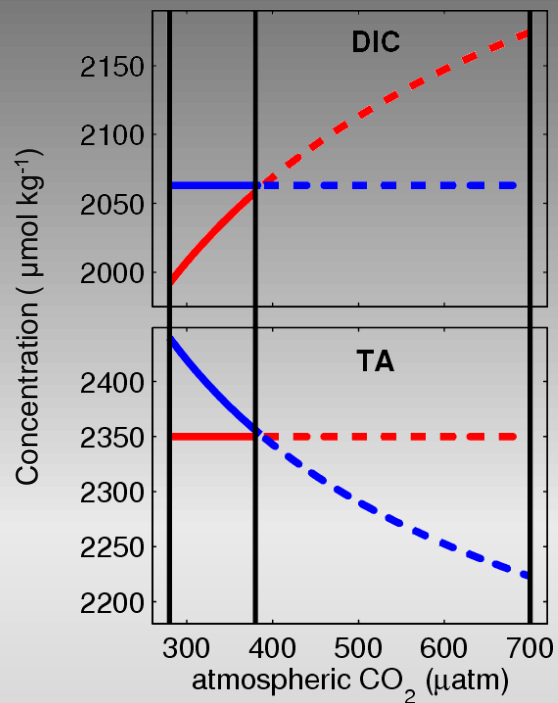
increasing DIC at constant TA

↑ [CO₂] , [HCO₃⁻]

↓ pH, [CO₃²⁻], Ω_{calc} , Ω_{arg}

Seawater carbonate redistributions also occur by variations in TA at constant DIC

Changes in DIC and TA



within the 280 -700 μatm range



quite similar variations
in concentration changes



only exception [HCO_3^-],
increasing $\sim 4\%$ vs. $\sim 9\%$
between 380 and 700 μatm

Different approaches for different experimental setups

- ➔ **Which organism / ecosystem is studied?**
autotrophs, heterotrophs, mixed communities
- ➔ **What is the size of the experiment?**
liters vs. cubic meters of seawater
- ➔ **What are the main processes impacting the carbonate system?**
photosynthesis vs. calcification or respiration
- ➔ **What is the duration of the experiment?**
hours/days vs. weeks
- ➔ **What is the main question?**
physiology vs. biogeochemistry



Leads ideally to a perfect way to manipulate, maintain and monitor the experimental carbonate system

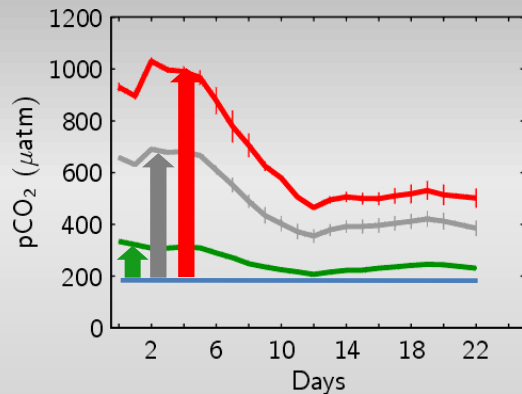
DIC manipulation : aeration of seawater at target CO₂ levels



HOWTO: aeration of the mesocosms (~30m³ each) with air enriched with CO₂ at target levels

aeration was stopped when levels were reached after about two days

Phytoplankton bloom was studied for about 3 weeks



CO₂ / pH change gradual



relatively easy to adjust although equilibration might take some time

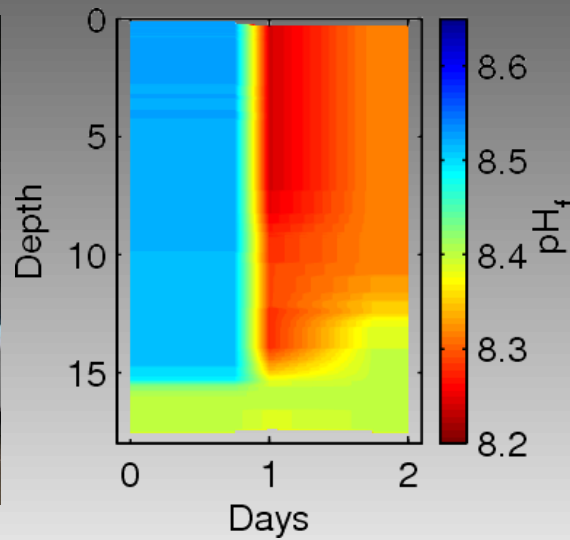


simultaneous supply of O₂ during aeration



direct seawater aeration might disturb organisms or impact dissolved organics pool

DIC manipulation : injection of CO₂ enriched water

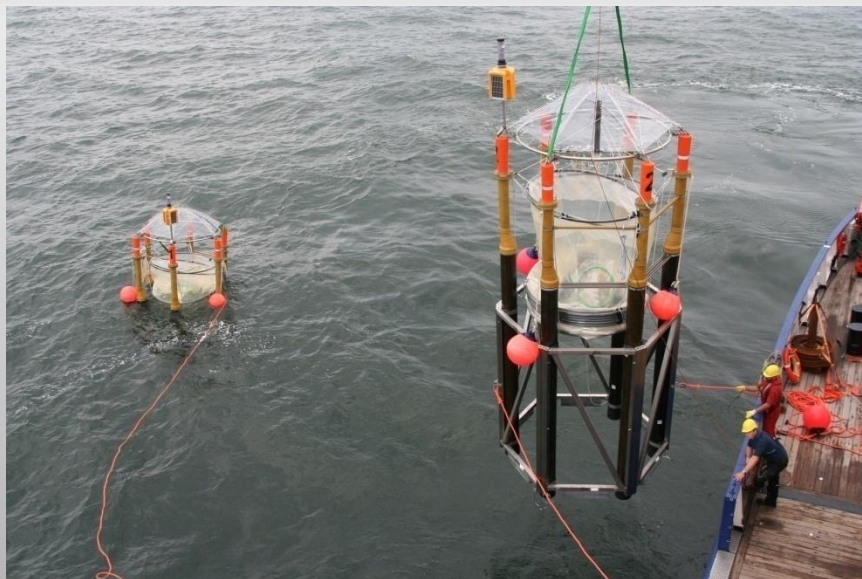


HOWTO:

eration of filtered seawater with pure CO₂

injection of CO₂ saturated seawater into mesocosms

about 100 l per 60m³ for
~0.3 pH drop



works extremely well for large volumes



does not require a lot of infrastructure



not so easy to adjust to desired CO₂ level

DIC manipulation : NaHCO_3 or Na_2CO_3 additions together with HCl



HOWTO: addition of NaHCO_3 to increase DIC to target values together with equimolar additions of HCl to counterbalance the otherwise increase of TA

additions of Na_2CO_3 with twice as much HCl



ideal for small scale bottle experiments

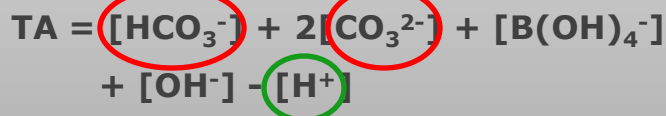


manipulations can be extremely precise up to a couple of micromoles.

dissolved inorganic carbon



total alkalinity (excluding some minor components)

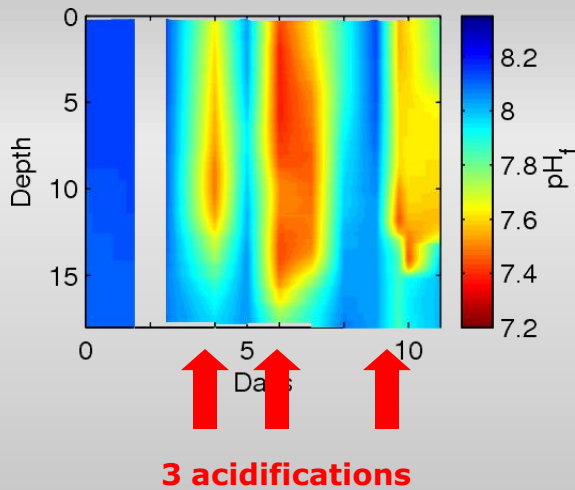


TA manipulation : NaOH / HCl additions



HOWTO:

increasing $p\text{CO}_2$ by additions of HCl and decreasing by NaOH



widely used, relatively easy way for manipulation



does not require any complicated infrastructure



feasible for small and large scale applications

Final remarks and summary

- ➔ **Type of manipulation depends on application, e.g. aeration at target $p\text{CO}_2$ might be necessary to supply O_2 for heterotrophs**
- ➔ **Manipulation is first step, equally important is monitoring**
- ➔ **Measure at least 2 parameters of the carbonate system to ensure that the manipulation has worked as intended**
- ➔ **Be sure to know how DIC and TA is impacted in the experiments as this will ultimately determine $p\text{CO}_2$ changes (long-term experiments With calcifiers at constant $p\text{CO}_2$)**
- ➔ **High nutrient (ammonium, silicate, phosphate) concentrations have to be considered as TA components**

Thank you

